Supplementary Information for

Non-Magnetic Bimetallic MOF-Derived Porous Carbon-Wrapped

TiO₂/ZrTiO₄ Composites for Efficient Electromagnetic Wave Absorption

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S1 Supplementary Animation



Animation 1 the surface current intensity of simulative $TiO_2/ZrTiO_4/carbon$ composites stimulated by electromagnetic waves

CST Microwave Studio is used for discretization and iterative solution of Maxwell equations for a given condition on the basis of time domain finite integration. A simplified model was set with the size and material parameters measured in this work, and the driven frequency is 10 GHz.



S2 Supplementary Tables and Figures



Experimental Evidences: To verify the statement that the phase discrepancy of TiO₂ would not affect the absorption performance, the experimental evidences were supplied as followed:

The titanium(IV) isopropoxide was calcined at 600, 700, and 800 °C in air atmosphere. Correspondingly, the XRD patterns (Fig. S2) indicated the anatase phase, mixed phase, and rutile phase TiO₂ were successfully prepared. The electromagnetic parameters of the different TiO₂ (filling rate 50 wt%) were measured (Fig. S3a–c), and the final electromagnetic wave absorption performances were calculated as well (Fig. S3d–f). The permittivity of the three samples was extremely close with each other. And all the samples exhibited similar weak absorption properties.



Fig. S2 PXRD patterns of anatase phase, mixed phase, and rutile phase TiO₂



Fig. S3 electromagnetic parameters of **a** anatase phase, **b** mixed phase, and **c** rutile phase TiO_2 ; and two-dimensional RL projection maps of **d** anatase phase, **e** mixed phase, and **f** rutile phase TiO_2



Fig. S4 TGA curves of all MOF derivatives in this work

The detailed calculation equation and analysis of component contents: For the $TiO_2/ZrTiO_4/carbon$ composites, the carbon content could be obtained first. In the precursors (PCN-415), the particular MOF crystal structures required the specific TiZr-oxo clusters ($[Ti_8Zr_2O_{12}(COO)_{16}]$). And the atomic ratios between Ti and Zr must be 4:1. Thus, the molar ratio between TiO₂ and ZrTiO₄ should be 3:1. Therefore, the weight ratio between TiO₂ and ZrTiO₄ should be $3M_{TiO2}:M_{ZrTiO4}$.

The specific calculation equations had been provided as followed. And no mistake could be found after careful checks.

$$wt\%_{\text{TiO}_2} = (1 - wt\%_{\text{Carbon}}) \times \frac{3M_{\text{TiO}_2}}{3M_{\text{TiO}_2} + M_{\text{ZrTiO}_4}}$$

S3 / S6



Fig. S5 XPS survey spectrum of TZC-7 and TZC-8



Fig. S6 FT-IR spectra of all MOF derivatives



Fig. S7 Simulation results by *CST Microwave Studio*: **a** model structure, **b** surface current intensity, and **c** inner surface current intensity



Fig. S8 ($|Z_{in} / Z_0|$) curves at 3.00 mm of matching thickness



Fig. S9 Theoretical matching thickness curves of **a** TZC-7, **b** TZC-8, and **c** TZC-9 drawn on the two-dimensional RL projection mappings



Fig. S10 Two-dimensional ($|Z_{in} / Z_0|$) value projection mappings of a **a** TC-7, **b** ZC-7, **c** TZC-6, **d** TZC-7, **e** TZC-8, and **f** TZC-9

Sample -	Atomic molar fraction (%)				Component weight fraction* (wt%)			
	СК	O K	Ti K	Zr L	carbon	TiO ₂	ZrTiO ₄	ZrO ₂
TZC-6	38.85	43.11	14.41	3.63	26.60	39.54	33.86	/
TZC-7	34.22	44.90	16.62	4.26	19.94	42.67	37.40	/
TZC-8	25.53	50.19	19.14	5.14	13.34	44.82	41.84	/
TZC-9	16.16	55.86	22.04	5.94	7.17	47.90	44.93	/
TC-7	31.61	49.17	19.22	/	26.42	73.58	/	/
ZC-7	32.27	50.98	/	16.75	24.43	/	/	75.57

Table S1 Atomic molar fractions characterized by EDS, and weight fraction of components

* The component weight fraction was calculated according to the atomic molar fraction.

Sample	Conductivity $(S \text{ cm}^{-1})$	Resistivity	Sheet Resistance (Ω/\Box)	Carrier Concentration	
TZC-6	1.32×10 ⁻⁷	7.60×10 ⁶	1.27×10 ¹¹	1.43×10 ¹¹	
TZC-7	3.70×10 ⁻¹	2.70×10 ⁰	4.51×10 ⁴	2.29×10 ¹⁹	
TZC-8	7.31×10^{0}	1.37×10^{-1}	2.28×10 ³	1.41×10^{19}	
TZC-9	1.70×10^{0}	5.89×10 ⁻¹	9.82×10 ³	2.16×10 ²¹	
TC-7	5.25×10 ⁻³	1.90×10^{2}	3.17×10^{6}	2.36×10 ¹⁴	
ZC-7	6.01×10^{-4}	1.66×10 ³	2.77×10^{7}	4.38×10 ¹⁴	

 Table S2 Conductive properties of all MOF derivatives in this work